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Biradical Formation from Molecules with (Z)-7-Sulfonyl-3-hexen-1,5-diyne Functionalities

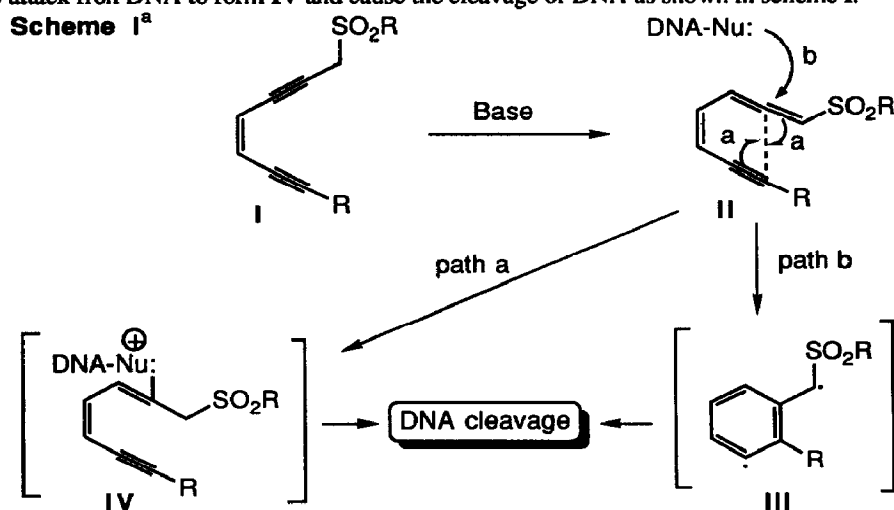
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Abstract: Molecule that contains (Z)-7-sulfonyl-3-hexen-1,5-diyne functionalities undergoes base-catalyzed isomerization to (Z)-ene-yne-allene-sulfone and subsequent Myers cyclization to form aromatic products, presumably via a biradical intermediate. In the presence of a nucleophile, (Z)-ene-yne-allene-sulfone served as a good Michael acceptor.

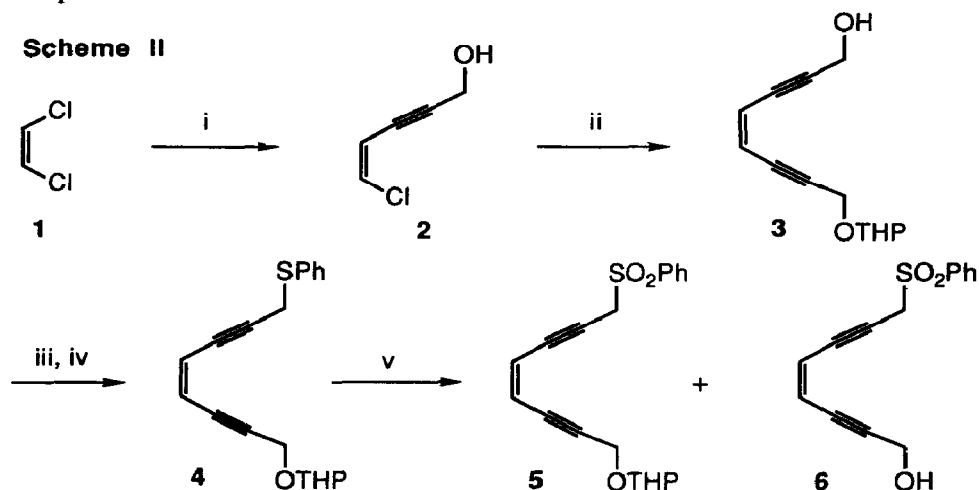
Among the recent studies on the enediyne antitumor antibiotics¹ related to neocarzinostatin,² esperamicin,³ calicheamicin,⁴ dynemicin,⁵ and most recent kedarcidin,⁶ molecules that contain the (Z)-allene-ene-yne functional group have been shown to undergo mild thermal reaction to form aromatic products⁷ and exist DNA-cleaving properties.⁸ Here, we demonstrate that a simple molecule that contains (Z)-7-sulfonyl-3-hexen-1,5-diyne substructure undergoes Myers cyclization reaction to form biradical under alkaline conditions.

The rationale design of this new class of compounds is based on the base-isomerization of propargyl sulfones to allenyl sulfones.⁹ Thus, compound I would be converted to allene-ene-yne II under an alkaline condition. Structure II was then expected to undergo either Myers cyclization reaction to form biradical III or nucleophilic attack from DNA to form IV and cause the cleavage of DNA as shown in scheme I.



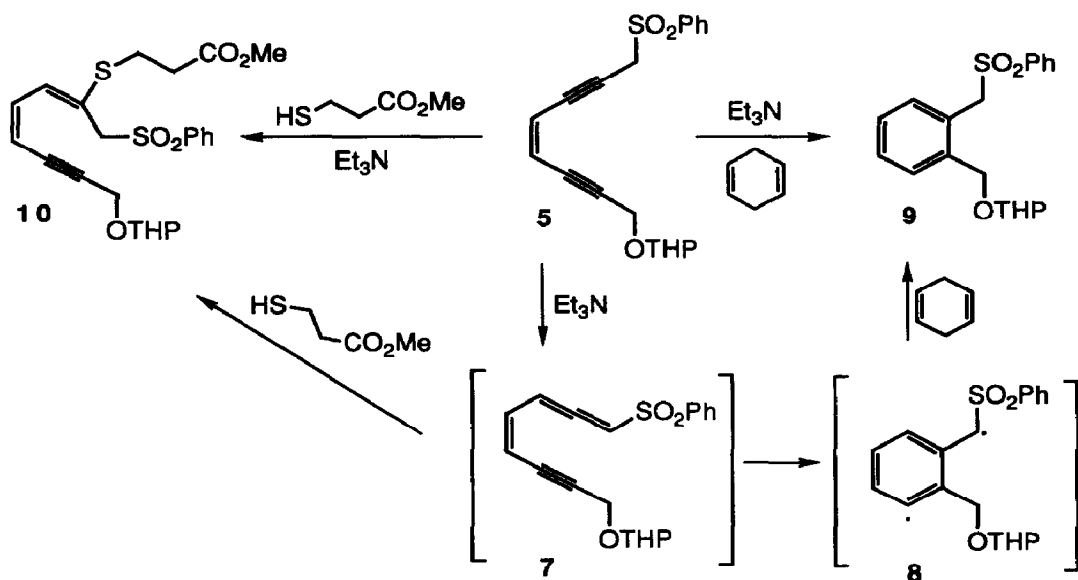
^aMechanistic rationale for ene-diyne-sulfones.

The synthesis of the representative compound **5** is outlined in scheme II. The synthesis of the key intermediate (Z)-enediyne **3** is readily achieved from commercial available (Z)-1,2-dichloroethylene **1** following the precedent of Myers.^{7c} Thus, reaction of propargyl alcohol (1 equiv) with (Z)-1,2-dichloroethylene (5 equiv), bis(triphenylphosphine) palladium (II) chloride (0.05 equiv) and cuprous iodide (0.15 equiv) in ethyl ether containing n-butylamine (5 equiv) at 30 °C for 6 h afforded, after isolation with flash column chromatography, the (Z)-vinyl chloride **2** in 65 % yield.^{7c} Subsequent coupling of **2** with protected propargyl alcohol (1.5 equiv) using tetrakis(triphenylphosphine)palladium (0) (0.05 equiv) as the catalyst under the same condition gave the (Z)-enediyne **3** in 50 % yield. Compound **3** was then converted into the corresponding mesylate by the standard method (MsCl, Et₃N in CH₂Cl₂). Subsequent reaction of the mesylate with thiophenol (1.0 equiv) under an alkaline condition (NaOH (1.0 equiv) in aqueous THF) afforded sulfide **4** in 44 % yield. Finally, oxidation of sulfide **4** with m-chloroperbenzoic acid (2.2 equiv) provided the sulfone **5** in 45 % yield along with 32 % of the deprotected product **6**.



reagents and conditions: i) HCCCH₂OH, Pd(PPh)₂Cl₂, Et₂O, CuI, BuNH₂, 65 %; ii) HCCCH₂OTHP, Pd(PPh₃)₄, Et₂O, CuI, BuNH₂, 50 %; iii) CH₃SO₂Cl, Et₃N, CH₂Cl₂; iv) HSPH, NaOH, THF-H₂O, 44 %; v) mCPBA, CH₂Cl₂, 45 % of **5** and 32 % of **6**.

In order to confirm whether the enediyne-sulfone **5** undergoes base-catalyzed conversion to eneyne-allene-sulfone **7** and subsequent cyclization to generate biradical **8**, the degassed solution of **5** in benzene (0.01 M) in the presence of 1,4-cyclohexadiene (1.5 M) was treated with Et₃N (5 equiv) at 30 °C for 10 h. After extractive isolation and flash column chromatography, the aromatized compound **9** was isolated in 45 % yield as a colorless oil. These results strongly suggested that the eneyne-allene-sulfone **7** and biradical intermediate **8** are actually involved in the transformation of enediyne **5** to aromatized compound **9**. On the other hand, the reaction of enediyne-sulfone **5** with methyl 3-mercaptopropionate (1 equiv) in the presence of Et₃N (5 equiv) in benzene (0.01 M) afforded the nucleophilic addition adduct **10**¹⁰ in 53 % yield. These results also suggested that eneyne-allene-sulfone **7** serves as an excellent Michael acceptor and possibly possesses DNA-cleaving properties.



In conclusion, molecules with (*Z*)-7-sulfonyl-3-hexen-1,5-diyne functionalities proceeded base-catalyzed conversion to (*Z*)-eneyne-allene-sulfones and subsequent Myers cyclization to form aromatized products or 1,2-addition reaction with nucleophiles. These results might provide an opportunity to development of a new class of DNA-cleaving antitumor agents.

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10. Some physical properties of **3**, **4**, **5**, **6**, **9**, and **10**: **3**: ^1H NMR (200 MHz, CDCl_3) δ 5.87 (s, 2H), 5.07 (t, 1 H, $J = 3.1$ Hz), 4.47 (d, 2 H, $J = 1.0$ Hz), 4.43 (d, 2 H, $J = 1.0$ Hz), 3.78-3.90 (m, 1 H), 3.55-3.64 (m, 1 H), 1.53-1.84 (m, 6 H); **4**: ^1H NMR (200 MHz, CDCl_3) δ 7.20-7.50 (m, 5 H), 5.82 (bs, 2 H), 4.88 (t, 1 H, $J = 3.1$ Hz), 4.42 (t, 2 H, $J = 1.1$ Hz), 3.80-3.89 (m, 3 H, including one set of doublet at 3.82 (d, 2 H, $J = 1.4$ Hz)), 3.51-3.59 (m, 2 H), 1.50-1.88 (m, 6 H); **5**: ^1H NMR (200 MHz, CDCl_3) δ 7.96-8.08 (m, 2 H), 7.55-7.74 (m, 3 H), 5.91 (dt, 1 H, $J = 11.0, 2.0$ Hz), 5.76 (dt, 1 H, $J = 11.0, 2.0$ Hz), 4.84 (t, 1 H, $J = 3.5$ Hz), 4.41-4.47 (m, 2 H), 4.17 (d, 2 H, $J = 2.0$ Hz), 3.78-3.90 (m, 1 H), 3.48-3.59 (m, 1 H), 1.49-1.88 (m, 6 H); **6**: ^1H NMR (200 MHz, CDCl_3) δ 8.03-8.09 (m, 2 H), 7.55-8.00 (m, 3 H), 5.96 (dt, 1 H, $J = 11.0, 2.0$ Hz), 5.76 (dt, 1 H, $J = 11.0, 2.0$ Hz), 4.46 (d, 2 H, $J = 2.0$ Hz), 4.17 (d, 2 H, $J = 2.0$ Hz), 2.16 (bs, 1 H); **9**: ^1H NMR (200 MHz, CDCl_3) δ 6.95-7.96 (m, 9 H), 4.31-4.70 (m, 5 H), 3.79-3.88 (m, 1 H), 3.48-3.57 (m, 1 H), 1.46-1.82 (m, 6 H); **10**: ^1H NMR (200 MHz, CDCl_3) δ 7.49-7.91 (m, 5 H), 6.65 (d, 1 H, $J = 11.4$ Hz), 6.38 (t, 1 H, $J = 11.4$ Hz), 5.40 (bd, 1 H, $J = 11.4$ Hz), 4.83 (bs, 1 H), 4.40-4.48 (m, 2 H), 4.10 (s, 2 H), 3.76-3.92 (m, 1 H), 3.69 (s, 3 H), 3.48-5.10 (m, 1 H), 3.00 (t, 2 H, $J = 7.0$ Hz), 2.58 (t, 2 H, $J = 7.0$ Hz), 1.39-1.90 (m, 6 H).

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